



The role of derivatization techniques in the analysis of plant cannabinoids by gas chromatography mass spectrometry

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ABSTRACT

Derivatization is the most powerful contribution to the identification and quantification of plant cannabinoids (p-CBDs) by gas chromatography-mass spectrometry (GC-MS): providing volatile derivatives with eminent properties (high selectivity, outstanding sensitivity and mass spectrometric peculiarities). These derivatives are excellent candidates to determine the main p-CBDs, like tetrahydrocannabivarin (THCV), tetrahydrocannabidivarin (CBDV), cannabidiol (CBD), cannabichromene (CBC), cannabicyclol (CBCL), tetrahydrocannabinol (THC), cannabigerol (CBG), cannabinol (CBN), 11-hydroxy- Δ^9 -THC (11-OH-THC) and 11-nor-9-carboxy-THC (THC-COOH). Identification and quantification of p-CBDs is required – partly as trace constituents in complex biological matrices of drug users, partly as main components in seizure samples: in both cases, in extremely different ratios. GC proposals published between 2000 and 2017, along with outstanding pioneer contributions, are reviewed. Procedures, without derivatization and applying various alkylsilyl, acylation and/or esterification techniques were listed, compared and criticized. Further sorting was based on the reagent type, on examined matrices, on enrichment/detection related acquisition protocols and on analytical performance characteristics.

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1. Introduction

The relevancy and continued interest towards chromatographic analyses of p-CBDs (called also marijuana) can be explained by their exponential spread in the community of drug users, started 6000 years ago [1] and continued by their partial legislation, recently [2]. Four states in the USA have legalized the cultivation, distribution and recreational use of marijuana; in addition the District of Columbia legalizes marijuana in some form. As to Europe, in the Netherland a hybrid system exists, reflecting Dutch policy [3]: in coffee shops namely, up to 5 g p-CBDs, in different forms are available; however plant's cultivation, trade and possession in the Netherlands is prohibited.

The continuously increasing popularity of p-CBDs in drug users' community is not questionable.

Literature overview of proposals – published in the time range of 2000 and 2017 [1–103] – reflects significant innovations to the field – compared to the pioneer works ([Supplemental files, pioneer proposals, Graphical Abstract: 1968 \[S1\]](#) [1966 \[S2\]](#) [1983 \[S3\]](#)).

The last comparison of GC-MS analytical techniques concerning p-CBDs' analysis was published in 1979 [S4].

It is worth mentioning that the pioneer methods dated back to the last third of the twentieth century including the activity of the pioneer 'green chemist' Rasmussen [S5, S6].

Out of reviews [4–11], appeared between 2000 and 2017 – specified exclusively to the p-CBDs [4,6,8,11] – were focused on their analysis in biological materials [4], on the efficacy of harm reduction strategy among people with schizophrenia [6], on legislation of driving under the influence of cannabis [8] and summing up micro extraction techniques for analysis of p-CBDs [11]. Reviews for illicit drugs in general, including also p-CBDs [5,7,9,10], were related to the solid phase micro extraction (SPME) techniques in analytical toxicology [5], to quantify p-CBDs in drug users' oral fluids [7], to enrich in micro-particles from the air [9], or, to perform dispersive liquid-liquid micro extraction (DLLME) in forensic toxicology [10].

The aim of this review was to pinpoint the advancements and advantages of GC analysis for the principal constituents of p-CBDs. The structure, physical chemical properties of p-CBDs, their possible reagents and the corresponding derivatized products are shown in Supplemental Files ([Table S1](#): structure of p-CBDs in initial forms; [Table S2](#): structure of derivatization reagents; [Table S3](#): structure of p-CBDs as silylated and acylated derivatives).

We summed up the relevant papers into four groups according to the techniques employed and to the analytes investigated:

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Table 1 shows methods developed for the analysis of p-CBDs in their initial forms, without derivatization [1,2,12–37], while **Tables 2–4** comprises methodologies suitable to identify and quantify compounds derivatized via alkylsilylation (**Tables 2 and 3**; [38–86]) and combined acylation/esterification techniques (**Table 4**; [88–103]).

Methods were further sub grouped in **Tables 1–4** according to the type of enrichment processes carried out from various, p-CBDs-containing matrices. Within such groupings, papers are listed in chronological order of their publication.

Moreover, a schematic overview of our approach is presented in **Fig. 1** that visualizes the matrices, methods and citations documenting p-CBDs' analysis by gas chromatography. Numerical data indicate that the most frequently analyzed samples vary from 28.7% (whole blood, plasma, or serum) to 1.1% (bile, airborne particles, meconium, e-Cigarette, Marijuana Formulation, fingernail), expressed in the total of our paper selection.

Criticality is focused on optimized derivatization conditions, reproducibility, selectivity, sensitivity indicated with analytical performance characteristics (LOD, LOQ values), as well as with time-, labor and complexity of methodologies.

2. GC of p-CBDs

In general, it is worth mentioning that derivatization does have a particular importance in the GC based protocols. This time consuming and tedious process, by many application chemists, preferring liquid chromatographic separation of underivatized species, was and still is regarded as the main disadvantage of sample preparation, needed prior to GC analyses. However, this so called 'disadvantage' is dwarfed in comparison to several advantages. This means that (i) on the one part GC methods provide increased selectivity, sensitivity and the possible identification and quantification of numerous species on a single column, simultaneously, while, (ii) on the other part they do not suffer of ion suppression phenomena as LC analyses of underivatized species do.

It is worthy to note that recently also in LC techniques – to be comparable with the GC-MS-MS standard protocol – derivatization was reported as a not avoidable step [98]. For this purpose, prior to the LC-MS-MS analysis of the THC-COOH accumulation in hair samples, an esterification step was inserted. The methyl ester of THC-COOH manifested increased stability, selectivity and sensitivity: corresponding in thorough accordance with those results

Table 1
Analysis of CBDs by gas chromatography (GC) without derivatization.

Matrix/amount	Enriched by	Acquisition	LOD ng/mL	LOQ ng/mg	Aim; compounds	Ref.
Leafy material/50 mg	C ₂ H ₅ OH extr	GC-FID	—	—	Chemotype selection; THC, CBD	[12]
Resinous floral blact/50 mg	CHCl ₃ extr	GC-FID	—	—	Chemotaxonomic analysis; THC, CBD, CBDV, THCV, CBC, CBGM, CBG	[13]
Grounded, dried marijuana/60 mg	HS-SPME 80–150°C	GC-MS	—	—	HS-SPME optimization (Swiss grown marijuana),THCV, CBCL, CBV, CBD, CBC; THC, CBG, CBN	[14,15]
Hair (3 mm cuts)/50 mg	*washing	GC-MS	THC: 414 CBN: 43.8	—	In vitro contamination of hair by marijuana smoke; THC, CBN	[16]
Soap bar resin/25 mg	CH ₃ OH	GC-MS	—	—	Cannabinoids' relative amounts definition; THC, CBN, CBD	[17]
Plasma/—	SPE	GC-MS-SIM	—	0.5–1.0	Time estimation of cannabis use; THC, THC-COOH	[18]
Hair (1 mm cuts)/10 mg	HS-SPME	GC-MS-SIM	0.07	0.12	Hair contamination study; THC, CBN, CBD	[19]
Powdered seizures/100 mg	CH ₃ OH/CHCl ₃ = 9/1 extr	GC-FID	3 µg/mg	5 µg/mg	Seizure analysis phenotype selection; THC, CBN, CBD	[20]
Seizure resin block/100 mg	CH ₃ OH extr	2D-GC-FID,TOF-MS	—	—	Pixel based chemometric analysis of seizures; THC, CBD, heroine	[21]
Saliva/0.2 mL	PMME**	GC-MS-SIM	0.68	—	Method optimization, THC	[22]
Cannabis/0.1 g	CH ₃ OH/CHCl ₃ = 9/1 extr	GC-FID	—	—	Propagation dependent distribution study of cannabinoids and seizure analysis; THC, THCV, CBD, CBC, CBG, CBN	[23–26]
Hair/10 mg	HS-SPME	GC-MS-MS	0.007–0.031	0.012–0.062	Method validation; THC, CBN, CBD	[27]
Hair/50 mg	LLE, pentane	GC-MS	0.01	0.02	Method development; THC	[28]
Milled plants/50 mg	SFE, FUSE	GC-MS	—	—	SFE, FUSE extractions' optimization/comparison; THC, CBD, CBN	[29]
Oral fluid/0.1 mL	SPME	GC-MS	0.5–20	2–690	Drug users' testing; THC, CBN, CBD	[30]
Plant material/0.2 g	CH ₃ OH extr	GC-MS	—	—	Cannabis profiling; THC, CBN, CBD, CBG	[31]
Plant/50 mg	SFE; SC-CO ₂	GCxGC-MS	—	—	Resolution study of <i>Cannabis sativa</i> CBDSS	[32]
Plant/25 g	C ₆ H ₆ extr	GC-MS	—	—	Cannabis profiling; THC (THC-COOH decarboxylated)	[33]
Powdered seizures/100 mg	CH ₃ OH/CHCl ₃ = 9/1 extr	GC-FID	—	—	Cannabinoids' seizure analysis; THC, THCV, CBD, CBC, CBG, CBN	[34]
Cannabis/50 mg	Vaporisers®	GC-MS-SIM	—	1–250 µg/mL	Smoke free inhalation optimization study; CBD, CBC, THC, CBN	[1]
Controlled liquid/1 mL	Tube dipping	DART-MS, GC-MS	—	—	Marijuana e-cigarette formulation study; CBDs and all and terpenes	[2]

Indications: extr = simple or ultrasound assisted shaking/extraction followed by filtration; BGM = cannabigerol monomethylether; *washing = exposed to marijuana smoke for 60 min, followed in three consecutive washing solutions, in total; SPE = solid phase extraction; **PMME = polymer monolith micro extraction; amount® = taken for one sample preparation; LOD/LOQ = limit of detection/limit of quantitation values; — = no data available; SPME = solid phase micro extraction; HS-SPME = headspace SPME; HS-SPDE = headspace solid phase dynamic extraction; LLE = liquid-liquid extraction; SFE = supercritical fluid extraction; SC-CO₂ = supercritical carbon dioxide; FUSE = focused ultrasound extraction; 2D = two dimensional; DART = Direct Analysis in Real Time. ng/mg throughout the LOD and LOQ values in Table has been described in italics.

Table 2

Analysis of cannabinoids, derivatized with BSTFA, determined by GC-MS.

Matrix/amount	Enriched by	Derivatized			Acquisition; m/z fragments	LOD ng/mL	LOQ ng/g	Aim; compounds	Ref
		With	°C	Min					
Blood, urine/1 mL	LLE, C ₆ H ₆ /EtAC = 9/1	BSTFA	70	15	GC-MS/MS; 372 → 305, 289	—	5–50	Analysis at pg level; THC-COOH	[38]
Urine/1.5 mL	SPE	BSTFA, 1% TMCS	70	10	GC-MS; 488, 473, 398, 371	2.44	9.48	Matrix optimization; THC-COOH	[39]
Plasma/1 mL	SPE	BSTFA, 1% TMCS	80	15	SPE with a single eluent; GC-MS-SIM(PCI); THC: 387, 11-OH-THC: 459, THC-COOH: 489	0.5–1.0	0.5–1.0	SPE simplification study; THC, 11-OH-THC, THC-COOH	[40]
Rabbit plasma/0.5 mL	SPE	1. BSTFA:PYR = 1:3 2. TFAA	60 Room temp	60	1. GC-MS-EI; THC-TMS: 386 2. GC-MS-NICI; THC-TFA: 410	— 0.3	10	Two steps derivatization, two ionization techniques; THC	[41]
Hair/50 mg (5 mm cut)	Washing: 1. neutral n; 2. acidic a,	1n. THC, CBN, CBD 2a BSTFA, 1% TMCS	60	20	GC-MS; 1n: without derivatization, 2a: as BSTFA derivative	n: 0.012 a: 0.024	n: 0.02 a: 0.080	Two fractions (n, a) analysis: 1. THC, CBD, CBN, 2.ThC-COOH	[42]
White pig blood/5 mL	LLE, C ₆ H ₆ /EtAC = 7/1	BSTFA, 1% TMCS	70	20	GC-MS; two fractions, separately silylated, determined	—	0.5–5	THC metabolism study/ animal model; THC → 11-OH-THC, THc-COOH	[43]
Plasma/1 mL	Hydr, SPE	BSTFA, 1% TMCS	80	45	GC-MS-SIM; THC: 387, 11-OH-THC: 459, THC-COOH: 492	—	0.5–1	Oral administration control; THC, 11-OH-THC, THc-COOH	[44]
Whole blood/1 mL	SPE; 1. n; 2. a,	BSTFA, 1% TMCS	70	20	2D-GC-MS; two fractions, eluted into the same vial	—	1.0	Derivatization optimization; THC, THc-COOH	[45]
Urine/1 mL	LLE, C ₆ H ₆ /EtAC = 7/1	BSTFA, 1% TMCS	60	30	GC-MS-SIM: 371, 374	—	20	Administered THC-COOH in urine	[46]
Plasma/1 mL	Hydr, SPE	BSTFA, 1% TMCS	70	30	2D-GC-MS, 1D: 15 m; criofocusing, 2D: 30 m)	0.125–0.25–0.125	m.o.; THC, 11-OH-THC, THc-COOH	[47]	
Hair/100 mg	HS-SPME-PDMS	BSTFA, 1% TMCS	125	20	GC-MS-SIM; THC: 303, 371, 386; CBD: 351, 390, 458; CBN: 310, 367, 382	0.01–0.02	0.39–4.2	m.o.; THC, CBD, CBN	[48]
Oral fluid/1 mL	SPE	BSTFA, 1% TMCS	60	15	GC-MS; THC: 386, 371, 303; CBN: 367, 382; THc-COOH: 487, 488	—	5.0	m.o.; THC, CBD, CBN, THc-COOH	[49]
Whole blood/1 mL, urine/2 mL	SPE	BSTFA, 1% TMCS	70	45	2D-GC-MS, 1D-GC-MS: 15 m; criofocusing, 2D: 30 m)	—	0.25–0.50	m.o [47], and abusers' urine control [48]; THC, 11-OH-THC, THc-COOH	[50,51]
Bile/1 mL	LLE; C ₆ H ₆ EtAC/ACA = 90:10:1	ACN/BSTFA, 1% TMCS = 1/1	70	30	GC-MS-SIM; THC: 303, 371, 386; THc-COOH: 371, 473, 488	0.28	0.86	m.o.; THC, THc-COOH	[60]
Urine/1 mL	HF-LPME	BSTFA, 1% TMCS	90	15	GC-MS-SIM	1.5	2.0	THC-COOH	[68]

Indications as in Table 1, as well as: (PCI) = positive chemical ionization; HF-LPME = hollow fiber-liquid phase micro extraction; m.o. = matrix and p-CBDs related method optimization; HS-SPME-PDMS = HS-SPME-polydimethylsiloxane; NICI = negative ion chemical ionization. ng/mg throughout the LOD and LOQ values in Table has been described in italics.

which have been obtained by the well established GC-NICI-MS-MS technique. This experience is a trend to increase selectivity and sensitivity also in the LC protocols.

2.1. Quantitation of p-CBDs in their initial forms, without derivatization (Table 1)

Concerning quantitation of plant matrices [13,23–26,34] flame ionization (FID) detection was regarded satisfactory providing excellent, fast results. Primarily GC-FID was applied when the ratios of the expected constituents had to be defined: based on the responses and retention properties of authentic standards.

Simultaneous analysis of seven [13] six [35] and five [23–26,34] p-CBDs were presented throughout the time reviewed, retaining the relevancy of this simple, cost-effective technique.

In addition, mass selective identification and quantification protocols were also employed mostly along with process validation [1,16,18,19,22,27,28,30,36,37].

2.2. Identification and quantification of p-CBDs as various derivatives (Tables 2–4)

Papers related to p-CBDs analysis in their initial forms, without derivatization (Table 1) undoubtedly confirmed that to identify and

quantify their trace amounts, especially in biological matrices, quantitative enrichment protocols, followed by labeling strategy is unavoidable and obligatory.

3. The alkylsilylation techniques

In the light of p-CBDs' hydroxyl and carboxyl functions (Table S1) to trigger and exhaust the unique challenge of alkylsilylation approaches is obvious; these protocols are primarily suitable to volatilize and improve mass fragmentation properties of active proton containing groups, simultaneously [38–86]. Out of the wide choice of alkylsilyl reagents, the most commonly used BSTFA [38–69] and MSTFA [70–86] were preferred. The use of MTBSTFA, due to its assumed steric hindrance, remained of marginal importance [78,87].

3.1. Derivatization with BSTFA (Table 2); method optimization and biological activity studies

BSTFA derivations are the most popular suggestions for p-CBDs analysis by GC.

Evaluating conditions and results compiled in Table 2, it reveals that all selected proposals do provide LOQ values: promising the quantitative aspect of studies.

Table 3

Analysis of cannabinoids, derivatized with MSTFA, determined by GC-MS.

Matrix/amount	Enriched by	Derivatized	Acquisition; <i>m/z</i> fragments			LOD ng/mL	LOQ ng/mg	Aim; compounds	Ref	
			With	°C	Min					
Hair/10 mg	HS-SPME, on fiber, fully automated	MSTFA		90	8	GC-MS-SIM; THC-TMS: 303, 371, 386; CBD-di-TMS: 301, 337, 390; CBN-TMS: 367, 368, 382	0.08–0.14	0.27–0.51	mo; CBD, CBN, THC	[70]
Hair/50 mg	Alkaline digestion, SPE	MSTFA/NH ₄ I/DTE = 100/2/4 (v/w/w)		70	30	GC-MS-MS; THC-TMS: 371; CBN-TMS: 367, CBD-TMS: 390; THC-COOH-TMS: 371	—	—	mo; THC, CBD, CBN, THC-COOH	[71]
Hemp product/1 mL or 1 g	LLE; C ₆ H ₆ /i-PrOH = 9/1	MSTFA, 0.2% TMCS		70	30	GC-MS-SIM, THC-TMS: 303, 371, 386; CBD-TMS 337, 390, 486; CBN-TMS: 310, 367, 382	0.3–0.6	0–2	mo; THC, CBD, CBN	[72]
Urine/2 mL	LLE; CHCl ₃ /iPrOH = 9/1	MSTFA, 1% TMCS		70	20	GC-MS-SIM, THC-COOH: 371, 473, 488	3	—	mo; THC-COOH	[73]
Fingernail/30 mg	basic/acidic hydr LLE; C ₆ H ₆ /ETAC = 9/1	MSTFA, 1% TMCS		70	15	GC-MS-SIM; THC-TMS: 315, 371, 386; THC-COOH-2TMS: 371, 473, 488	<0.056–0.2	— mo; THC, THC-COOH + A, MA + MDMA)	mo; THC-COOH	[74]
Urine/2 mL	SPE*	MSTFA/NH ₄ I/DTE (500:4:2, v/w/w)		60	20	GC-MS-SIM; THC: 386; OH-THC and THC-COOH: 371, 386	0.1	0.2	mo; THC, OH-THC, THC-COOH	[75]
Rat urine/1 mL	enzymatic hydr; SPE	MSTFA/ETAC = 1/1		90	45	GC-MS-SIM	—	—	metabolism study: THC-COOH to THC	[76]
River/500 mL, waste water/100–200 mL	SPE	MSTFA		80	60	GC-MS-MS; THC-TMS: 386 → 315, 330; THC-2TMS: 473 → 355	0.9–1	2.7–3.0	THC, THC-COOH + 10 others	[77]
Urine/5 mL	enzymatic hydr; LLE; C ₆ H ₆ /ETAC = 7/1	MSTFA		80	30	GC-MS; CBG-2TMS: 337, 391, 377, 460	—	—	mo; metabolism study: CBG	[78]
Blood (post mortem)/1 mL	LLE (C ₆ H ₆ : ETAC = 5/1, 2×), centrifuged	MSTFA		70	60	2D-GC-MS-SIM, CBD: 390; THC: 371; CBN: 367; 11-OH-THC: 371; THC-COOH: 371	0.25	0.25–0.50	CBD, THC, CBN, 11-OH-THC, THC-COOH	[79]
Hair/50–100 mg	LLE, 1 basic: THC; 2. acidic: THC-COOH	1.MSTFA, 0.2% TMIS 2. PFPAA/PFPOH		60	20	GC-MS/MS, (NCI)	0.01 pg/mg	0.04 pg/mg	two fractions analysis: THC, THC-COOH	[80]
Waste water/10 mL	SPME (60°C, 60 min)	MSTFA (on fiber)		70	—	GC-MS	1–2.5	3.3–8.3	THC, THC-COOH	[81]
Urine/1 mL	LLE	MSTFA/ethanethiol/NH ₄ I (500:4:2), MW:750 W, 1.5 min.		40	10	GC-MS/MS(SRM); 371 → 305, 371 → 289, 371 → 265, 371 → 95	0.057	0.19	mo; THC-COOH	[82]
Blood (post mortem)/1 mL	SPE	MSTFA		80	30	GC-MS; THC-2TMS: 315, 371, 386; THC-COOH-2TMS: 371, 474, 488	5	10	mo; THC, THC-COOH, (cocaine + amphetamines)	[83]
Plasma or serum/1 mL	SPE	MSTFA		70	20	GC-MS-SIM; THC: 303, 371, 386; OH-THC: 371, 459, 474; THC-COOH: 297, 371, 473	0.15–2.0	0.30–3.30	mo; THC, 11-OH-THC, THC-COOH	[85]
Serum/1 mL	LLE fully automated	MSTFA		80	30	GC-MS-SIM; THC: 303, 371, 386; OH-THC: 371, 459, 474; THC-COOH: 371, 473, 488	0.2–0.6	0.6–2.3	mo; THC, 11-OH-THC, THC-COOH	[86]

Indications as in Tables 1 and 2, as well as/or: *SPE = performed with calcium-hardened β-cyclodextrin polymer; DTE = dithioerytrol; i-PrOH = isopropyl alcohol.
ng/mg throughout the LOD and LOQ values in Table has been described in italics.

Table 4
Analysis of cannabinoids, derivatized with acylation and/or esterification, determined by GC-MS.

Matrix/amount	Enriched by	Derivatized	Acquisition; <i>m/z</i> fragments			LOD ng/mL	LOQ ng/mg	Aim; compounds	Ref
			With	°C	Min				
Blood/1 mL	SPE	TFAA/CHCl ₃ = 1/1	70	10	GC-MS-SIM; THC: 410, 11-OH-THC: 408	—	0.5	mo; THC, 11-OH-THC	[88]
Hair/50 mg	extr. CHCl ₃ /i-PrOH	PFPAA/PFPOH	65	30	GC-MS; CBD: 377, THC: 377, THC-COOH-2PFP: 489, CBD: 231, CBN: 295	—	—	mo; CBD, CBN, THC, THC-COOH	[89]
Blood/1 mL	extr. CH ₆ H ₆	PFPAA/PFPOH = 2/1	70	25	GC-MS-SIM; THC: 417, 445, 460	0.5	—	mo; THC	[90]
Serum/1 mL	SPE	1.TBAH/DMSO = 5/1 2. CH ₄ I,	Room temp → 2 → 10	GC-MS, THC: 313, 328, 245, 11-OH-THC: 313, 314, 358, THC-COOH: 313, 357, 372	0–49–0.65	0.62–3.35	mo; THC, 11-OH-THC, THC-COOH	[91]	
Blood, serum/25 µL	SPE	TMAH/DMSO = 1:50, v/v, CH ₃ I	Room temp	15	GC-MS-SIM, THC: 313, 328, 11-OH-THC: 313, 358, THC-COOH: 313, 357	0.7–0.8	—	mo; THC, 11-OH-THC, THC-COOH	[92]
Sweat patches	extr. CH ₃ OH + SPE	TFAA (TEA)*	80	20	GC-MS(NICI); THC: 410	0.2 ng/p	0.4 ng/p	mo; THC	[93]
Blood/—	SPE	CHCl ₃ /TFAA/HFIPOH = 2/2/1	70	25	GC-MS-SIM; THC: 410, THC-COOH: 422	—	2.5	time estimation of cannabis use; THC, THC-COOH	[94]
Hair/50 mg	hydr → extr. C ₇ H ₇ /ETAC = 9/1	PFPAA/PFPOH = 5/3	70	30	GC-MS-MS(NICI); THC-COOH: 513 → 293, 363, 470	50 pg/mg	100 pg/mg	mo; THC-COOH	[95]
Hair/25 mg	hydr → extr. CH ₆ H ₆ /ETAC = 9/1	PFPAA/PFPOH = 2/1	70	30	GC-MS/MS(NICI); 611 → 483	0.02 pg/mg	0.05 pg/mg	mo; THC-COOH	[96]
Blood/0.5 mL	LLE: CH ₆ H ₆ /ETAC = 9/1	TFAA/HFIP = 2/1	70	25	GC-MS-MS(NICI), SRM; THC: 410 → 313; THC-OH: 409 → 339; THC-COOH: 422 → 361	0.1–0.02	0.5–2.5	mo; THC, THC-OH, THC-COOH	[97]
Hair/20 mg	hydr → LLE extr. C ₆ H ₆ /ETAC = 9/1	PFPAA/PFPOH = 2/1	70	30	GC-MS-MS(NICI), SRM; THC: 459 → 3973; THC-COOH: 602 → 474	2.5–25	7.5–50	mo; THC, THC-COOH	[98]
Hair/25 mg (mechanically pulverized)	Hydr → LLE extr. C ₆ H ₆ /ETAC = 9/1	PFPAA/PFPOH = 5/3	70	30	GC-MS-MS(NICI), SRM; THC-COOH: 602 → 474, 602 → 513	0.015 pg/mg	0.05 pg/mg	mo; THC-COOH	[99]
Urine/0.5 mL	LLE, 1. basic: As 2. acidic: THC-COOH	1. TFAA 2. PFPOH	50 530	45 530	GC-MS-SIM; THC-COOH: 572	0.86	2.88	mo; THC-COOH + amphetamines	[100]
Hair/50–100 mg	MeOH extr	PFPAA/HFIPOH	—	—	GC-MS-MS/(NICI), 620 → 383, 492	—	0.1 pg/mg	mo; comparison to LC-MS-MS; THC-COOH	[101]
Oral fluid/1 mL	SPE	TFAA/HFIP = 2/1	65	40	GC-MS-MS; THC-COOH: 522 → 490	0.0075	0.0010	mo; THC-COOH	[102]
Urine/1 mL	NaOH hydr → acidified → extr: C ₆ H ₆ /ETAC = 5/1	PFPAA/PF = 1/1	75	30	GC-MS-SIM; THC-COOH: 473, 459, 607, 622	1	2	THC-COOH elimination study from urine; THC-COOH	[103]

Indications as in Tables 1–3, as well a/or: TBAH/DMSO = tertabutylammonium hydroxide/dimethylsulfoxide reagent; CH₄I = iodomethane derivatization reagent; TMAH = tetramethylammonium hydroxide; HFIPOH = hexafluoroisopropanol; TFAA (TEA)*100 µL of 0.01 mol/L triethylamine in heptane, and 20 µL of TFAA; p = patch.
ng/mg throughout the LOD and LOQ values in Table has been described in italics.

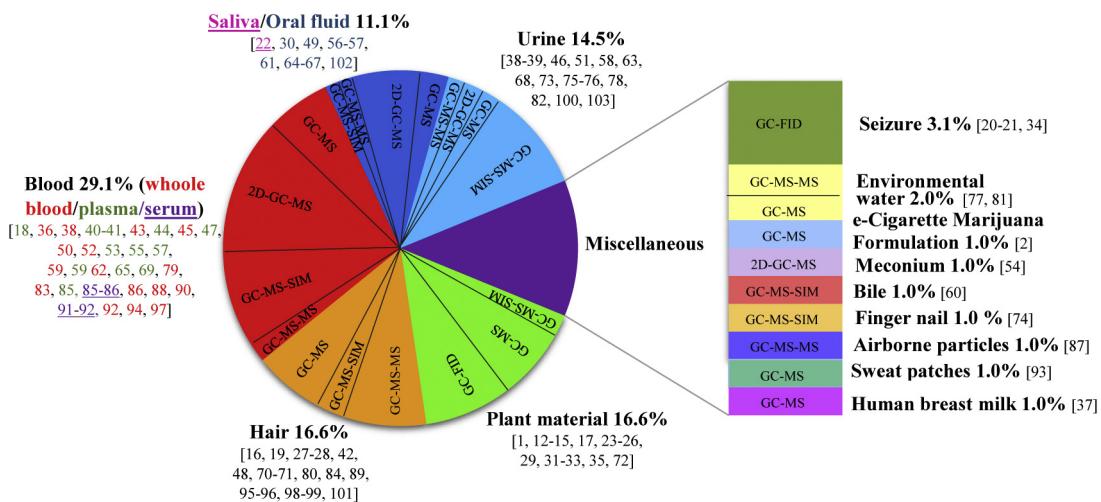


Fig. 1. Matrix, method and citations related analysis of cannabinoids by gas chromatography. Note: Literature overview of proposals was selected on Web of Science, Science Direct and Scopus basis obtained with the key words of Cannabinoids, gas chromatography in the publication time range of 2000–2017, including the novelty containing ones. Excluding papers that (i) did not give detailed sample preparation protocol, simply referred to a previously published article, as well as (iii) related to synthetic-, and/or endo-CBDs.

With two exceptions [41,42] analyses were carried out in one fraction. Independently of the species to be labeled, the use of 1% trimethylchlorosilane (TMCS) containing BSTFA was preferred. Also, BSTFA of its own [38], or in dilutions with pyridine (PYR) [41] was applied. Conditions for temperature (60–125°C) and time (10–60 min) of derivatizations were varied: consequently in the frame of this review optimized derivatization conditions can not be suggested.

Due to the intrinsic peculiarities of microextractions they are only suitable to determine the neutral [48] and acidic species [68], separately. GC-MS-SIM [40,44,47,48,60,68]; and GC-MS-MS acquisition was performed in a single case, only [38].

It is worth to note that this type of method developments regarding basic researches and innovative practical proposals equally, are significantly associated with the activities of NIH experts, leaded by Huestis [40,44,47,50–59,61–67,69]. It means, in addition to the details listed in Table 2, additional studies [61–67,69] manifested increased method selectivity performed via two dimensional (2D) GC-MS. This concept was extended and many sided utilized. Because of their relevant content, considered to be interesting in the community of analysts, biochemists and toxicologists, these are summed up briefly.

- Providing a scientific database to asses p-CBDs in oral fluids (OF) [61].
- Characterizing p-CBD elimination from blood of daily cannabis smokers by highlighting the usefulness of improving the accuracy of results interpretation [62].
- Optimizing the alkaline and enzymatic hydrolysis conditions to the highest p-CBDs recovery in urine [63].
- Following THC concentrations under therapy: meaning 4–5 days of monitored abstinence from smoking [64].
- Evaluating the relationship between OF and plasma p-CBDs concentration depending on dose, route of administration and time after dosing [65].
- Studying OF collection device impact on p-CBDs stability [66]. Since, reliable analytical and technical conditions were of primary importance in specifying time courses of THC, 11-OH-THC, THC-COOH, CBN and CBD concentrations during multiple and ad libitum smoking periods [67].
- Defining pharmacokinetics of p-CBDs in plasma after controlled and ad libitum smoking making possible differentiation between chronic/frequent and single users [69].

3.2. Derivatizations with MSTFA (Table 3)

MSTFA, due to its stronger silyl donor behavior – next to BSTFA, out of seventeen cases in seven – was used without TMCS and solvents [67,77–79,83,86]. In two cases MSTFA was completed with 0.2% [72] or 1% [73,74] TMCS. Forced conditions were ensured applying the MSTFA 0.2% TMIS [80], or MSTFA/NH₄I/DTE [71,75], or MSTFA/C₂H₅SH/NH₄I [82] reagents. As solvent, in 1/1 vol ratios, ethyl acetate [76,84] was performed. Unfortunately the advantages of the use of special catalysts and solvents were not detailed.

Regarding the analytical performance characteristics – we have to accept that the comparisons after various enrichment working strategies, performed from different matrices are not, or moderately comparable. The optimum method selection is a matter of the technical and personal preparedness of the laboratory where the task is to be solved.

3.3. Derivatization via N-tert-butyldimethylsilyl-N-methyltrifluoroacetamide (MTBSTFA)

A relevant topic was introduced to determine THC, CBD and (CBN), the primary active constituents of cannabis preparation, in airborne particulates [87]. The specific procedure consists of soot extraction by ultrasonic bath, purification by solvent partitioning, derivatization with MTBSTFA, and analysis by applying an optimized tandem GC-MS-MS technique. The proposal proved to be suitable to quantitate the three psychotropic substances at concentrations ranging from ~0.001 to ~5.0 ng/cm air. The procedure was performed on field in Rome and Bari (Italy), demonstrating that all three compounds contaminate the air in Italian cities. The comparison of MTBSTFA with MSTFA in the analysis of THC and 11-OH-THC revealed the preference of trimethylsilylation [81].

4. Acylation and/or esterification methods (Table 4)

All proposals in this section are matrix and method related optimization studies taking into consideration in particular the quantitation of THC-COOH.

Analyte enrichment processes are similar to detailed before (Tables 2 and 3).

Characteristic, combined derivatizations consist of acylation and esterification [89–92,94–103]. In our evaluation all these processes are overcomplicated; mostly aiming the full derivation of THC-

COOH. Two exceptions [88,93], in favor to quantify 11-OH-THC and THC [88] or THC [93] exclusively, acylation was performed.

These two reagents applying techniques, based on separate reactions with the hydroxyl and carboxyl moieties with one exception [93] were suggested in the analysis of biological matrices.

5. Comparison of derivatization approaches (Tables 2–4)

Reaction temperature and time of reagent depending techniques are not comparable: from these points of view even within the same reagent, basic researches fail. Especially as suggested reaction times and temperatures are varying inexplicably. BSTFA and MSTFA were applied in wide temperature and time ranges: BSTFA (Table 2) between 60°C for 15 min [49] and 125°C, for 20 min [48], while MSTFA (Table 3) between 40°C for 10 min [81] and 90°C, for 45 min [76].

Acquisition techniques' types were applied prominently in different ratios comparing the BSTFA, MSTFA and acylation/esterification derivatizations. Beside the simple GC-MS protocols [89,91,93], both the GC-MS-SIM [88,90,92,94,100,103] and the GC-MS-MS [95–100] techniques were represented. GC-MS-MS protocols, in increasing order of listing, in 20% (BSTFA: 3/15 × 100), in 40% and 40% (in MSTFA and in acylation/esterification, cases, equally: 6/15 × 100), respectively.

As to the analytical performance characteristics – due to the intrinsic peculiarities of the task to be solved – derivatization techniques' unambiguous comparison is impossible. Notwithstanding, contrasting the same sample type and size (blood/serum/urine, 0.5–2 mL), applying the GC-MS-SIM [47,75,103] and the tandem GC-MS-MS [38,82,97] acquisition protocols, LOQ values proved to be as follows:

GC-MS-MS: 5 ng/mL [38], 0.19 ng/mL [82] and 0.5 ng/mL [97];

GC-MS-SIM: 2.0 ng/mL [46], 0.2 ng/mL [75], 2.0 ng/mL [103];

In the light of these values MSTFA derivatization might be preferred.

Authors of this review are convinced that the use of two reagents in order to obtain full derivatization of p-CBDs and/or to perform analysis in two fractions is needless: (i) resulting in complexity of processes and (ii) leading to increased time and work consume. (iii) In addition, two fractions' analysis involves a potentially increased loss of the analyte.

6. Final remarks and future trends

Accordingly the aim and scope of this review specific attention was paid to consider sample related optimized conditions, and to define the necessary future round of the duties to point out contradictory literature suggestions.

Summing up experiences of GC proposals for p-cannabinoids' analysis – published in the time range of 2000–2017 – in order of importance, it can be stated that

- a) gas chromatography retained its relevancy as standard separation technique.
- b) In cases of plant matrices, including seizures, FID detection of underderivatized species – even at the present – proved to be satisfactory. Excellent, fast results can be obtained primarily when the ratios of the constituents had to be defined, only. In these cases, due to the availability of plenty of materials, analytical performance characteristics were of marginal importance. Simultaneous GC-FID analysis of seven p-CBDs was presented throughout the time reviewed, saving the timeliness of this simple, cost-effective technique.
- c) The importance of derivatization approaches providing outstanding mass spectrometric properties along with increased selectivity and sensitivity is not questionable. Out

of the most common alkylsilylation and acylation/esterification processes, in authors understanding, alkylsilylation labeling is the methods of choice, because hydroxyl and carboxyl moieties are reacting at once.

- d) Acquisition techniques were applied prominently in different ratios comparing the BSTFA, the MSTFA and the acylation/esterification processes. Beside the simple GC-MS protocols, the GC-MS-SIM and the GC-MS-MS techniques were contrasted. Distribution of these protocols, indicating in total of papers commented, shows up in 20% (BSTFA: 3/15 × 100) and in 40% (both MSTFA and acylation/esterification: 6/15 × 100), respectively.
- e) In order to designate derivatization/acquisition techniques of choice it is impossible: an approximate comparison has been provided. As a possible approach, considering the similar sample type and size (blood/serum/urine, 0.5–2 mL) the LOQ values – obtained in the GC-MS-SIM and the tandem GC-MS-MS acquisition protocols of variously derivatized species – have been contrasted. In this correlation the following LOQ values can be compared:

GC-MS-MS: 5 ng/mL (BSTFA), 0.19 ng/mL (MSTFA) and 0.5 ng/mL (acylation/esterification);

GC-MS-SIM: 2.0 ng/mL (BSTFA), 0.2 ng/mL (MSTFA) and 2.0 ng/mL (acylation/esterification);

- f) On the basis of this literature overview and self experiences – because of their many-sided suitability and unique efficiency – we are convinced on the general advantages of trimethylsilyl processes: introduced by Pierce [104], followed by thousands of researchers and confirmed also in our earlier approaches [105–108] associated with the analysis of various, active proton containing species.
- g) As an endpoint of this review authors state that
 - in cases of plant matrices [13,23–26,34,35] flame ionization (FID) detection was regarded satisfactory providing excellent, fast results. Primarily GC-FID was applied when the ratios of the expected constituents had to be defined. Simultaneous analysis of seven [13], six [35] and five [23–26,34] p-CBDs were presented throughout the time reviewed, saving the relevancy of this simple, cost-effective technique.
 - In contrary when selectivity, sensitivity, reliability and reproducibility are of primary importance, (infinitive amounts of species are to be isolated, identified and quantified from biological tissues) independent of the type of matrix, trialkylsilylation and tandem mass spectrometry should be the method of choice.
- h) As future perspective authors of this review are convinced that alkylsilyl derivatization optimization should be performed in the same laboratory, on the same apparatus to be comparable. The p-CBDs' trialkylsilylation related, optimized working strategy, using GC-MS and/or GC-MS-MS, in authors' laboratory is in progress.

Appendix A. Supplementary data

Supplementary data related to this article can be found at <http://dx.doi.org/10.1016/j.trac.2017.07.022>.

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